

# UNIDIRECTIONAL SEEDED SINGLE CRYSTAL GROWTH FROM SOLUTION OF SULPHAMIC ACID AND ITS CHARACTERIZATION

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## **ABSTRACT**

Unidirectional seeded single crystal growth by SR Method. The growth crystal has a cylindrical morphology with good optical quality by this method. The entire solute can be converted in crystal so that, 100 % solute crystal efficiency can be achieved. The huge sulphamic acid crystal grown from a seed crystal in a super saturated aqueous solution. In the instrument a vertically designed “L” bend was used to avoid the spurious nucleation. Growth from solution is often the only alternative if the substance decomposes below its melting point or undergoes a phase change between its melting point and temperature at which crystal is to be used. The growth rate at which perfect

crystal can be grown from solution are appreciably smaller than in process of growth from melt. The linear rates of growth from solution are commonly 10mm per day compare to rates of 10mm per minute from the melt. In SR method orientation seed was fixed at the bottom of the glass and ampoule. And crystal of diameter 40 mm & length 60mm successfully grown by SR method.

**Key word** :- Growth from solution, XRD, FTIR, Single Crystal Growth, Inorganic compound

## **1. INTRODUCTION**

Now day nonlinear optical materials are widely used in many different areas of solid state physics. In the recent past organic non linear optics (NLO) material are gaining attention because of their good nonlinear properties over inorganic counter parts. The organic nonlinear optical (NLO) crystal plays an important role in second harmonic generation (SHG), frequency mixing, Electro optic modulation and optical parametric oscillation. Crystallization from the solution is an important process and is a two step process nucleation and crystal growth. The driving force for crystallization is the degree of super saturation which has

been commonly expressed as the difference in concentration between the super saturated and saturated solution. Sulphamic acid is a classical inorganic compound. It is an important and industrial chemical with an annual manufacturer of several kilotons. It is moderately soluble in water. It is stable at room temperature and normal atmosphere. The unidirectional seeded single crystal is a promising NLO material. A novel growth and S.R. Method introduce for growth of single crystal from aqueous solution. This technique already applied to grow some organic and inorganic crystal. Sankarnarayan method is better than conventionally growth crystal. The goal is to be combining the potential high polarizability of organic molecules such as aromatic cycle, with thermal, mechanical and chemical stability of mineral compound. This method is important and extremely flexible for the development of new material applicable over a very board range of frequency. The advantage is of semi organic is that, the crystal can grow from aqueous solution and they form three dimensional crystals which can be easily cut and polished. Good quality and large size nonlinear optical crystal are need for laser application. For practical

application, we need good optical transparency and also crystal should with stand high optical power and should have chemical stability. The grown crystal was characterized by X – Ray powder diffraction analysis, FTIR study etc.

## 2. EXPERIMENTAL SETUP

The schematic diagram of S.R. method apparatus is shown in fig.1. It consists of growth ampoule made out of glass with seed mounting pad. The diameter of the growth portion and top portion of the ampoule was 20mm & 80mm respectively. A ring heater kept at the top of the growth ampoule & bottom of the growth ampoule was connected to the temperature. The temperature around growth ampoule kept constant with the help of temperature controller till the process is completed. The grown crystal fig.2.

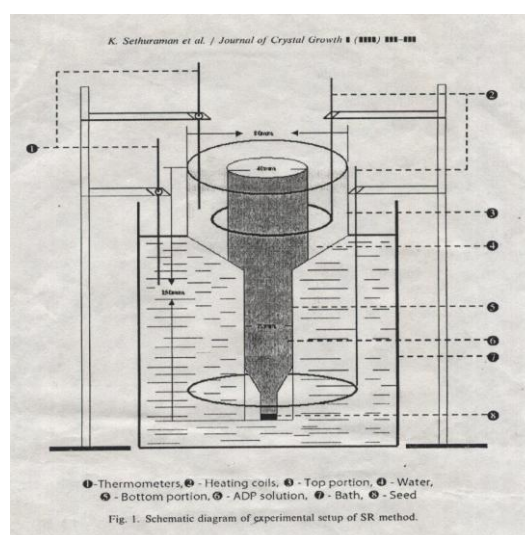


Fig. 1 Schematic diagram of S.R. Method Apparatus.



Fig. 2 Single crystal of S.A.

The ring heater was moved downwards using a translation mechanism designed “L” bend, the growth condition of this method depend on the heating point.

The solubility experiment was carried out in constant temperature bath for the temperature 30° C, 35° C, 45° C & 55° C.

The growth experiment was performed using demonized water and mixed solvent with the help of solubility curve by the slow evaporation technique. The temperature of solution was increase slowly. The color of the solution becomes slightly yellowish after a few days at temperature above 60° C so that temperature of solution was kept always below 60° C.

#### **Characterization: a) FTIR Analysis**

The functional group was identified by FTIR using JASCO model 610 FTIR spectrometer in the range 6000 to 2000  $\text{cm}^{-1}$  as shown in fig.3. The presence of band between 4000- $^{-1}$  to 3800  $\text{cm}^{-1}$  are mainly due to N-H stretching. The band

observe at 2400  $\text{cm}^{-1}$  arises symmetric vibration of  $\text{NH}_3^+$  group and the band at 4000  $\text{cm}^{-1}$  is an a symmetric stretching. The strong band 5600  $\text{cm}^{-1}$  correspond to  $\text{SO}_3$  stretching vibration. This peak was also observed at FTIR spectrum. It always shows that, reflectance range of active band 3000 to 500  $\text{cm}^{-1}$ . The band observed with the wave number of 1400  $\text{cm}^{-1}$  in reflectance is goes unsymmetrical stretching vibration. From the FTIR analysis of the compound of sulphamic acid was identified. The band observed 1200 to 1300  $\text{cm}^{-1}$  & strong band also found 1350 & 1450  $\text{cm}^{-1}$  as shown in fig.4.

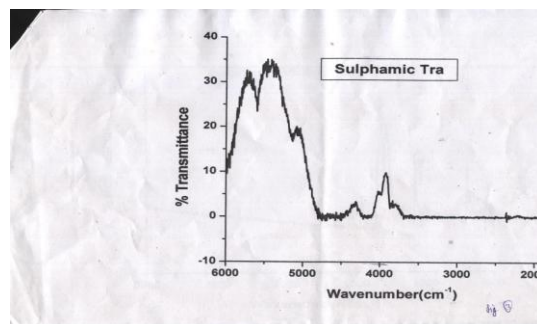


Fig. 3 FTIR spectrum of S. A.

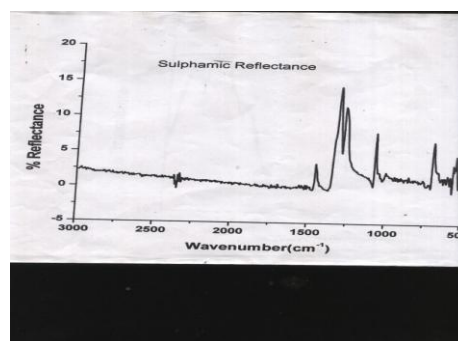


Fig. 4 FTIR spectrum of sulphamic acid

#### **b) X-Ray powder Diffraction Analysis**

The peak observed from fig.5, X – Ray diffraction spectrum were analyzed & indexed using software package and the data obtained by the powder XRD analysis are in accordance with the single seeded crystal XRD data ,shown in the data spectrum table. The single crystal XRD reveals that the sulphamic acid crystal belongs to orthorhombic structure.

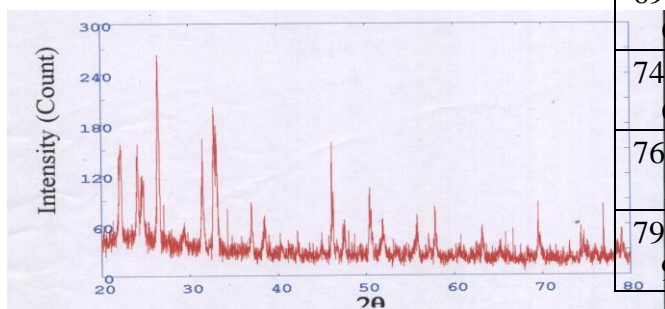


Fig. 5 Powder X-Ray Diffraction pattern of sulphamic acid

0	38			
46.14	1.965	132.5	1017.0	0.2400
8	45			
47.52	1.911	32.7	586.9	0.3800
50.41	1.808	64.4	899.7	0.2800
6	63			
51.91	1.759	34.1	544.7	0.3200
5	85			
55.80	1.646	31.8	492.2	0.3600
3	09			
57.84	1.592	45.3	569.2	0.2600
5	75			
63.14	1.471	26.3	439.2	0.3400
6	20			
69.53	1.350	56.7	566.0	0.2200
5	80			
74.66	1.270	27.2	271.5	0.2000
6	19			
76.94	1.238	25.7	282.0	0.2200
1	20			
79.01	1.210	34.2	341.7	0.2000
9	77			

XRD spectrum Table

2Thet a	d (A)	Heig ht	Area	FWH M
22.19 4	4.002 08	105.8	2008. 6	0.3800
24.11 0	3.688 24	104.3	1341. 5	0.2600
24.63 3	3.611 18	74.5	1234. 9	0.3600
26.49 1	3.361 89	209.0	2851. 0	0.2800
29.37 9	3.037 69	23.7	426.7	0.3600
31.45 7	2.841 59	100.8	1308. 1	0.2600
32.76 0	2.731 46	155.5	1659. 4	0.2200
36.99 7	2.427 85	54.8	685.0	0.2600
38.45	2.339	44.9	655.4	0.3000

### **RESULT AT A GLANCE:**

Sulphamic acid seeded single crystal was successfully grown by slow evaporation solution grown technique. The solubility of sulphamic acid in demonized water at saturated solution the FTIR and X–Ray diffraction studies confirmed the vibration frequency of sulphamic acid. The crystal also found to be NLO material having good quality in transparent medium.

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### **REFERENCE**

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- 5 C. R. Khothari  
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- 6 K. Nassau, E. Schoheer, C. G. Pro & Apal  
Berly Heidberg New York.
- 7 S.J. Hicking R.J. Wooley, lecturer of  
Physics
- 8 N.E.Hill W.E. Vughan A.H. Price and M.  
Davies College of London
- 9 K. Lark Hortvitz, Vivin A, Johnson  
Department of physics (U.S.A.)
- 10 K. Sethuraman, R. Rameshbabu, R. Gopa  
Ikrishnan, P. Ramasamy Journal of  
Crystal Growth I (III) III-III

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